| C24 | $-0.3665(2)$ | $-0.1059(2)$ | $-0.0931(4)$ | $0.0673(9)$ |
| :--- | :--- | :---: | :---: | :--- |
| C25 | $-0.3621(2)$ | $-0.2250(2)$ | $-0.2097(3)$ | $0.0683(10)$ |
| O1W | $0.42272(15)$ | $0.17651(14)$ | $0.5381(2)$ | $0.0772(7)$ |

Table 2. Selected geometric parameters ( $\AA,{ }^{\circ}$ )

| $\mathrm{N} 1-\mathrm{C} 6$ | $1.331(3)$ | $\mathrm{C} 12-\mathrm{C} 13$ | $1.482(3)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{N} 1-\mathrm{C} 2$ | $1.352(3)$ | $\mathrm{C} 13-\mathrm{N} 14$ | $1.295(3)$ |
| $\mathrm{C} 2-\mathrm{C} 3$ | $1.377(4)$ | $\mathrm{N} 14-\mathrm{N} 15$ | $1.401(3)$ |
| $\mathrm{C} 3-\mathrm{C} 4$ | $1.387(4)$ | $\mathrm{N} 15-\mathrm{C} 16$ | $1.295(3)$ |
| $\mathrm{C} 4-\mathrm{C} 5$ | $1.388(3)$ | $\mathrm{C} 16-\mathrm{C} 17$ | $1.448(4)$ |
| $\mathrm{C} 5-\mathrm{C} 6$ | $1.411(3)$ | $\mathrm{C} 17-\mathrm{C} 18$ | $1.387(3)$ |
| $\mathrm{C} 5-\mathrm{C} 13$ | $1.460(3)$ | $\mathrm{C} 17-\mathrm{C} 22$ | $1.401(3)$ |
| $\mathrm{C} 6-\mathrm{C} 7$ | $1.476(3)$ | $\mathrm{C} 18-\mathrm{C} 19$ | $1.366(4)$ |
| $\mathrm{C} 7-\mathrm{N} 8$ | $1.349(3)$ | $\mathrm{C} 19-\mathrm{C} 20$ | $1.415(3)$ |
| $\mathrm{C} 7-\mathrm{C} 12$ | $1.393(3)$ | $\mathrm{C} 20-\mathrm{N} 23$ | $1.368(3)$ |
| $\mathrm{N} 8-\mathrm{C} 9$ | $1.346(3)$ | $\mathrm{C} 20-\mathrm{C} 21$ | $1.402(3)$ |
| $\mathrm{C} 9-\mathrm{C} 10$ | $1.372(4)$ | $\mathrm{C} 21-\mathrm{C} 22$ | $1.371(4)$ |
| $\mathrm{C} 10-\mathrm{C} 11$ | $1.387(4)$ | $\mathrm{N} 23-\mathrm{C} 24$ | $1.442(4)$ |
| $\mathrm{C} 11-\mathrm{C} 12$ | $1.370(3)$ | $\mathrm{N} 23-\mathrm{C} 25$ | $1.451(4)$ |
| $\mathrm{C} 6-\mathrm{N} 1-\mathrm{C} 25$ | $114.6(2)$ | $\mathrm{C} 11-\mathrm{C} 12-\mathrm{C} 7$ | $119.8(2)$ |
| $\mathrm{N} 1-\mathrm{C} 2-\mathrm{C} 3$ | $124.6(3)$ | $\mathrm{C} 11-\mathrm{C} 12-\mathrm{C} 13$ | $131.9(2)$ |
| $\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4$ | $120.0(3)$ | $\mathrm{C} 7-\mathrm{C} 12-\mathrm{C} 13$ | $108.3(2)$ |
| $\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 5$ | $117.2(2)$ | $\mathrm{N} 14-\mathrm{C} 13-\mathrm{C} 5$ | $133.9(2)$ |
| $\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 6$ | $118.2(2)$ | $\mathrm{N} 14-\mathrm{C} 13-\mathrm{C} 12$ | $119.6(2)$ |
| $\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 13$ | $133.2(2)$ | $\mathrm{C} 5-\mathrm{C} 13-\mathrm{C} 12$ | $106.4(2)$ |
| $\mathrm{C} 6-\mathrm{C} 5-\mathrm{C} 13$ | $108.6(2)$ | $\mathrm{C} 13-\mathrm{N} 14-\mathrm{N} 15$ | $113.7(2)$ |
| $\mathrm{N} 1-\mathrm{C} 6-\mathrm{C} 5$ | $125.4(2)$ | $\mathrm{C} 16-\mathrm{N} 15-\mathrm{N} 14$ | $110.0(2)$ |
| $\mathrm{N} 1-\mathrm{C} 6-\mathrm{C} 7$ | $126.7(2)$ | $\mathrm{N} 15-\mathrm{C} 16-\mathrm{C} 17$ | $122.7(3)$ |
| $\mathrm{C} 5-\mathrm{C} 6-\mathrm{C} 7$ | $107.9(2)$ | $\mathrm{C} 18-\mathrm{C} 17-\mathrm{C} 16$ | $123.4(2)$ |
| $\mathrm{N} 8-\mathrm{C} 7-\mathrm{C} 12$ | $124.5(2)$ | $\mathrm{C} 22-\mathrm{C} 17-\mathrm{C} 16$ | $119.3(2)$ |
| $\mathrm{N} 8-\mathrm{C} 7-\mathrm{C} 6$ | $126.8(2)$ | $\mathrm{N} 23-\mathrm{C} 20-\mathrm{C} 21$ | $121.8(2)$ |
| $\mathrm{C} 12-\mathrm{C} 7-\mathrm{C} 6$ | $108.7(2)$ | $\mathrm{N} 23-\mathrm{C} 20-\mathrm{C} 19$ | $121.4(2)$ |
| $\mathrm{C} 9-\mathrm{N} 8-\mathrm{C} 7$ | $113.9(2)$ | $\mathrm{C} 20-\mathrm{N} 23-\mathrm{C} 24$ | $121.2(2)$ |
| $\mathrm{N} 8-\mathrm{C} 9-\mathrm{C} 10$ | $125.4(3)$ | $\mathrm{C} 20-\mathrm{N} 23-\mathrm{C} 25$ | $121.2(3)$ |
| $\mathrm{C} 9-\mathrm{C} 10-\mathrm{C} 11$ | $119.6(3)$ | $\mathrm{C} 24-\mathrm{N} 23-\mathrm{C} 25$ | $117.2(3)$ |
| $\mathrm{C} 12-\mathrm{C} 11-\mathrm{C} 10$ | $116.9(3)$ |  |  |

All the H atoms were located from difference Fourier maps and refined isotropically.

Data collection: XSCANS (Siemens, 1994). Cell refinement: XSCANS. Data reduction: XSCANS. Program(s) used to solve structure: SIR92 (Altomare et al., 1994). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: ORTEPII (Johnson, 1976), SHELXTL/PC (Sheldrick, 1990). Software used to prepare material for publication: SHELXL93. Geometric calculations: PARST (Nardelli, 1983).

The authors would like to thank the Malaysian Government and the Universiti Sains Malaysia for research grant R\&D No. 123-3417-2201, and the State Science and Technology Commission, National Nature Science Foundation of China, for a grant for a Key Research Project. One of the authors (KS) thanks the Universiti Sains Malaysia for a Visiting Post Doctoral Research Fellowship.

[^0]
## References

Altomare, A., Cascarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. \& Camalli, M. (1994). J. Appl. Cryst. 27, 435436.

Baxter, P. N. W., Connor, J. A., Wallis, J. D., Povey, D. C. \& Powell, A. K. (1992). J. Chem. Soc. Perkin Trans. 1, pp. 1601-1605.

Chemla, D. S. \& Zyss, J. (1987). Editors. Nonlinear Optical Properties of Organic Molecules and Crystals. Vols I \& II. Orlando: Academic Press.
Fun, H.-K., Sivakumar, K., Zhu, D.-R. \& You, X.-Z. (1995). Acta Cryst. C51, 2076-2078.
Houlton, A., Jasim, N., Roberts, R. M. G., Silver, J., Cunningham, D., McArdle, P. \& Higgins, T. (1992). J. Chem. Soc. Dalton Trans. pp. 2235-2241.
Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
Nardelli, M. (1983). Comput. Chem. 7, 95-98.
Sheldrick, G. M. (1990). SHELXTL/PC Users Manual. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
Sheldrick, G. M. (1993). SHELXL93. Program for the Refinement of Crystal Structures. Univ. of Göttingen, Germany.
Siemens (1994). XSCANS. X-ray Single Crystal Analysis System. Version 2.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
Velavan, R., Sivakumar, K. \& Anbu, M. (1995). Acta Cryst. C51, 1227-1229

Acta Cryst. (1995). C51, 2080-2083

# Benzyl 3-(4-Dimethylaminobenzylidene)dithiocarbazate 

Hoong-Kun Fun, Kandasamy Sivakumar $\dagger$ and<br>Boon-Chuan Yip

X-ray Crystallography Laboratory, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia

Yu-Peng Tlan, Chun-Ying Duan, Zhong-Lin Lu and Xiao-Zeng You

Coordination Chemistry Institute, State Key Laboratory of Coordination Chemistry, Centre for Advanced Studies in Science and Technology of Microstructures, Nanjing University, Nanjing 210093, People's Republic of China
(Received 2 February 1995; accepted 29 March 1995)


#### Abstract

The title Schiff base compound, $\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{~S}_{2}$, crystallizes in the triclinic system with two molecules in the asymmetric unit. The dimethylaminophenyl and carbazate parts of the molecule lie in one plane, perpendicular to the plane of the $S$-benzyl group. One of the two molecules in the asymmetric unit displays lateral disorder of the $S$-benzyl group. The molecular packing is

^[ $\dagger$ On leave from: Department of Physics, Anna University, Madras 600 025, India. ]


governed by $\mathrm{N}-\mathrm{H} \cdots \mathrm{S}$ hydrogen bonds leading to the formation of dimeric pairs of the same type, i.e. pairs of either normal or disordered molecules.

## Comment

In recent years, there has been considerable interest in the chemistry of Schiff base compounds containing N and S donors. The increased activity in this field may be because of their medicinal properties and their chelating ability with transition metal ions (Ali \& Tarafdar, 1977; Ali \& Bose, 1984; Davies, El-Sayed, El-Toukhy \& Henary, 1990; Xu, Alyea, Ferguson \& Jennings, 1991). In this paper, we report the structure of a new Schiff base ligand derived from benzyl dithiocarbazate, namely, benzyl 3-(4-dimethylaminobenzylidene)dithiocarbazate, (I).


Fig. 1 shows an ORTEP (Johnson, 1965) plot of molecule 1 and a perspective view of molecule 2 , with the respective numbering schemes. Of the two molecules in the asymmetric unit, molecule 1 is normal, whereas molecule 2 is disordered. Bond lengths and angles in these molecules are generally normal. The $\mathrm{N} 1-\mathrm{Cl}$ single-bond distance is shorter than usual; 1.324 (3) for molecule 1 and $1.332(4) \AA$ for 2 . There are some differences in the $\mathrm{C} 1-\mathrm{S} 2$ and $\mathrm{S} 2-\mathrm{C} 11$ bond lengths of the disordered portion of molecule 2 (Table 2). The dimethylaminophenyl and dithiocarbazate portions of the molecule lie in one plane, due to the conjugated nature of the system. The $S$-benzyl group, however, is twisted at right angles relative to this plane. The dihedral angles between the phenyl, amino, carbazate and $S$-benzyl planes are given in Table 3.

In the crystal lattice the normal and disordered molecules exist as dimers of the same type, linked by $\mathrm{N}-\mathrm{H} \cdots \mathrm{S}$ hydrogen bonds $\left[\mathrm{H} 1 \cdots \mathrm{~S} 1^{\mathrm{i}} 2.60(2), \mathrm{N} 1 \cdots \mathrm{~S} 1^{i}\right.$ 3.411 (2) $\AA, \mathrm{N} 1-\mathrm{H} 1 \cdots \mathrm{~S}^{i}{ }^{i} 157(2)^{\circ}$ for molecule 1 ; $\mathrm{N} 1 A-\mathrm{S} 1 A^{\mathrm{iii}} 2.56$ (4), $\mathrm{H} 1 A \cdots \mathrm{~S} 1 A^{\mathrm{ii}} 3.396$ (4) $\AA, \mathrm{N} 1 A-$ H1A $\cdots$ S1 $A^{\text {ii }} 166(3)^{\circ}$ for molecule 2 ; symmetry codes: (i) $1-x,-y,-z$; (ii) $1-x, 1-y, 2-z$ ]. The molecules connected by these hydrogen bonds are related by a centre of symmetry. It is also observed that the molecules are stacked as ( $22 \overline{2}$ ) planar layers containing alternating normal and disordered molecules, the dihedral angle between the planes being $12.6^{\circ}$. An interesting feature is that the molecular orientations in these two planes are rotated by almost $90^{\circ}$ with respect to each other, as can be seen in Fig. 2. The inversion centres of the two

(a)

(b)

Fig. 1. (a) ORTEPII (Johnson, 1976) plot of molecule 1, shown with $50 \%$ displacement ellipsoids, and (b) a perspective view of molecule 2 , showing the disorder.


Fig. 2. Plot showing a projection of two layers of molecules. The disordered molecules are shown with open bonds and $\mathrm{N}-\mathrm{H} \cdots \mathrm{S}$ hydrogen bonds are shown as dashed lines.
groups in Fig. 2 lie on the $b$ axis, but with respect to the planar stacking, these inversion centres are related by a translation, which leads to the positioning of the $S$-benzyl group of molecule 2 sandwiched between inversion centres of molecule 1 . Hence, we attribute the
disorder in the $S$-benzyl group of molecule 2 to the N - $\mathrm{H} \cdots \mathrm{S}$ hydrogen bonds surrounding it. It may also be noted that the disorder in the $S$-benzyl group is only lateral, parallel to its own plane and perpendicular to the plane formed by the $\mathrm{N}, \mathrm{H}$ and S atoms which are involved in the hydrogen bonds. Since the $S$-benzyl group of molecule 1 is well away from the N and S atoms of molecule 2, it does not undergo any disorder. Among the two disordered $S$-benzyl sets, group $\mathrm{S} 2 B$ to $\mathrm{C} 17 B$ has almost the same conformation as that of the normal molecule 1, whereas its counterpart, S2A to C17A, deviates to a greater extent (relevent torsion angles are given in Table 2).

## Experimental

The title compound was prepared by refluxing benzyl dithiocarbazate and dimethylaminobenzaldehyde in a 1:1 molar ratio in methanol. The product was a yellow crystalline powder. Single crystals suitable for X-ray diffraction analysis were grown from benzene at room temperature.

## Crystal data

$\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{~S}_{2}$
$M_{r}=329.47$
Triclinic
$P \overline{1}$
$a=10.431(2) \AA$
$b=13.477(2) \AA$
$c=15.016(4) \AA$
$\alpha=109.56(2)^{\circ}$
$\beta=101.56(2)^{\circ}$
$\gamma=109.14(1)^{\circ}$
$V=1761.4(6) \AA^{3}$
$Z=4$
$D_{x}=1.242 \mathrm{Mg} \mathrm{m}^{-3}$
Data collection
Siemens P4 diffractometer
$\theta / 2 \theta$ scans
Absorption correction: none
6297 measured reflections
5284 independent reflections 3765 observed reflections
$[I>2 \sigma(I)]$
$R_{\text {int }}=0.0140$

## Refinement

Refinement on $F^{2}$
$R(F)=0.0464$
$w R\left(F^{2}\right)=0.1382$
$S=1.049$
5284 reflections
446 parameters
Only H -atom $U$ 's refined

$$
\begin{aligned}
& w= 1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.0729 P)^{2}\right. \\
&+0.2318 P] \\
& \text { where } P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3
\end{aligned}
$$

Mo $K \alpha$ radiation
$\lambda=0.71069 \AA$
Cell parameters from 25
reflections
$\theta=8-20^{\circ}$
$\mu=0.302 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Prism
$0.48 \times 0.46 \times 0.32 \mathrm{~mm}$
Pale yellow

$$
\begin{aligned}
& \mathrm{N} 1 \\
& \mathrm{~N} 2
\end{aligned}
$$

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters $\left(\AA^{2}\right)$

| $U_{\text {eq }}=(1 / 3) \sum_{i} \Sigma_{j} U_{i j} a_{i}^{*} a_{j}^{*} \mathbf{a}_{i} \cdot \mathbf{a}_{j}$. |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $x$ | $y$ | 2 | $U_{\text {eq }}$ |
| S1 | 0.45849 (9) | 0.13434 (8) | 0.10013 (5) | 0.0860 (3) |
| N1 | 0.6102 (2) | 0.0281 (2) | 0.1481 (2) | 0.0696 (6) |
| N2 | 0.7086 (2) | 0.0112 (2) | 0.21178 (15) | 0.0668 (6) |
| N3 | 1.0959 (3) | -0.2328 (2) | 0.3579 (2) | 0.0859 (7) |
| C1 | 0.5805 (3) | 0.1195 (2) | 0.1793 (2) | 0.0653 (7) |
| C2 | 0.7239 (3) | -0.0812 (2) | 0.1692 (2) | 0.0675 (7) |
| C3 | 0.8200 (3) | -0.1175 (2) | 0.2202 (2) | 0.0609 (6) |
| C4 | 0.8374 (3) | -0.2143 (2) | 0.1651 (2) | 0.0735 (7) |
| C5 | 0.9278 (3) | -0.2525 (2) | 0.2092 (2) | 0.0756 (8) |
| C6 | 1.0048 (3) | -0.1958 (2) | 0.3125 (2) | 0.0654 (7) |
| C7 | 0.9864 (3) | -0.0981 (2) | 0.3684 (2) | 0.0676 (7) |
| C8 | 0.8970 (3) | -0.0608 (2) | 0.3231 (2) | 0.0655 (7) |
| C9 | 1.1074 (4) | -0.3371 (3) | 0.2998 (3) | 0.1114 (12) |
| C10 | 1.1600 (4) | -0.1823 (3) | 0.4662 (2) | 0.0962 (10) |
| S2 | 0.67764 (8) | 0.21738 (6) | 0.30542 (5) | 0.0772 (2) |
| C11 | 0.6108 (4) | 0.3286 (3) | 0.3209 (2) | 0.0918 (9) |
| C12 | 0.6910 (3) | 0.4214 (2) | 0.4268 (2) | 0.0728 (7) |
| C13 | 0.8261 (4) | 0.5084 (3) | 0.4548 (3) | 0.0958 (10) |
| C14 | 0.8979 (4) | 0.5926 (3) | 0.5555 (4) | 0.1112 (13) |
| C15 | 0.8345 (6) | 0.5878 (4) | 0.6245 (3) | 0.1150 (14) |
| C16 | 0.7033 (6) | 0.5048 (4) | 0.5982 (3) | 0.1201 (13) |
| C17 | 0.6327 (4) | 0.4226 (3) | 0.5003 (3) | 0.0991 (10) |
| S1A | 0.62739 (13) | 0.44610 (8) | 1.09140 (7) | 0.1125 (4) |
| N1A | 0.5321 (3) | 0.3690 (2) | 0.8953 (2) | 0.0863 (7) |
| N2A | 0.5211 (3) | 0.3051 (2) | 0.7983 (2) | 0.0806 (7) |
| N3A | 0.2900 (3) | 0.0614 (2) | 0.3121 (2) | 0.0878 (7) |
| C1A | 0.6157 (3) | 0.3689 (3) | 0.9750 (2) | 0.0865 (9) |
| C2A | 0.4311 (3) | 0.3099 (2) | 0.7301 (2) | 0.0774 (8) |
| C3A | 0.3995 (3) | 0.2465 (2) | 0.6236 (2) | 0.0693 (7) |
| C4A | 0.3020 (3) | 0.2582 (3) | 0.5540 (2) | 0.0760 (7) |
| C5A | 0.2658 (3) | 0.1987 (3) | 0.4527 (2) | 0.0768 (8) |
| C6A | 0.3245 (3) | 0.1219 (2) | 0.4137 (2) | 0.0693 (7) |
| C7A | 0.4228 (3) | 0.1099 (2) | 0.4837 (2) | 0.0764 (8) |
| C8A | 0.4587 (3) | 0.1711 (2) | 0.5849 (2) | 0.0763 (8) |
| C9A | 0.1878 (4) | 0.0747 (4) | 0.2411 (2) | 0.1093 (12) |
| C10A | 0.3329 (4) | -0.0309 (3) | 0.2730 (2) | 0.1045 (11) |
| S2A | 0.6822 (3) | 0.2626 (3) | 0.9388 (4) | 0.1076 (14) |
| C11A | 0.7795 (7) | 0.2535 (8) | 1.0510 (6) | 0.097 (2) |
| C12A | 0.9309 (6) | 0.2756 (9) | 1.0649 (8) | 0.061 (3) |
| C13A | 1.0263 (12) | 0.3866 (8) | 1.0841 (8) | 0.103 (3) |
| C14A | 1.1731 (10) | 0.4148 (9) | 1.1047 (7) | 0.161 (7) |
| C15A | 1.2244 (6) | 0.3322 (14) | 1.1062 (10) | 0.169 (6) |
| C16A | 1.1290 (12) | 0.2212 (12) | 1.0870 (9) | 0.147 (4) |
| C17A | 0.9822 (10) | 0.1929 (7) | 1.0664 (6) | 0.098 (3) |
| S $2 B$ | 0.7342 (3) | 0.3066 (3) | 0.9481 (3) | 0.0943 (10) |
| C11B | 0.8305 (8) | 0.3472 (7) | 1.0789 (4) | 0.096 (2) |
| C12B | 0.9471 (7) | 0.3089 (9) | 1.0779 (9) | 0.067 (3) |
| C138 | 1.0876 (10) | 0.3844 (7) | 1.0962 (8) | 0.095 (3) |
| C14B | 1.1959 (6) | 0.3456 (9) | 1.0994 (10) | 0.146 (5) |
| C15B | 1.1636 (10) | 0.2313 (11) | 1.0843 (10) | 0.144 (4) |
| C16B | 1.0231 (12) | 0.1559 (7) | 1.0661 (6) | 0.125 (4) |
| C17B | 0.9149 (7) | 0.1946 (8) | 1.0629 (6) | 0.094 (3) |

Table 2. Selected geometric parameters $\left(\AA,^{\circ}\right)$

| $(\Delta / \sigma)_{\max }=0.001$ | $\mathrm{~S} 1-\mathrm{C} 1$ |
| :--- | :--- |
| $\Delta \rho_{\max }=0.323 \mathrm{e} \AA^{-3}$ | $\mathrm{~N} 1-\mathrm{C} 1$ |
| $\Delta \rho_{\min }=-0.378 \mathrm{e} \AA^{-3}$ | $\mathrm{~N} 1-\mathrm{N} 2$ |
| Atomic scattering factors | $\mathrm{N} 2-\mathrm{C} 2$ |
| from International Tables | $\mathrm{N} 3-\mathrm{C} 3$ |
| for Crystallography (1992, | $\mathrm{N} 3-\mathrm{C} 9$ |
| Vol. C, Tables 4.2.6.8 and | $\mathrm{C} 2-\mathrm{S} 2$ |
|  | $\mathrm{~S} 2-\mathrm{C} 11$ |
| 6.1.1.4) | $\mathrm{C} 11-\mathrm{C} 12$ |
|  | $\mathrm{~S} 1 A-\mathrm{C} 1 A$ |
|  | $\mathrm{~N} 1 A-\mathrm{C} 1 A$ |

$1.671(3)$
$1.324(3)$
$1.389(3)$
$1.274(3)$
$1.377(3)$
$1.441(4)$
$1.443(4)$
$1.742(3)$
$1.448(4)$
$1.817(3)$
$1.498(4)$
$1.664(3)$
$1.332(4)$
$\mathrm{N} 1 A-$
$\mathrm{N} 2 A$
$\mathrm{~N} 3 A$
$\mathrm{~N} 3 A$
$\mathrm{~N} 3 A$
$\mathrm{C} 1 A$
$\mathrm{C} 1 A$
$\mathrm{C} 2 A$
$\mathrm{~S} 2 A$
$\mathrm{C} 11 A$
$\mathrm{~S} 2 B$
C 11
1.381 (3) 1.277 (4) 1.370 (4)
1.440 (4)
1.451 (4)
$1.451(4)$
$1.746(4)$
1.763 (4)
1.444 (4)
$1.444(4)$
$1.850(8)$
1.465 (8)
1.814 (7)
1.470 (7)


Table 3. Dihedral angles $\left({ }^{\circ}\right)$ between the planes of various molecular groups

|  | Molecule 1 | Molecule $1 A$ | Molecule $2 B$ |
| :--- | :---: | :---: | :---: |
| Phenyl/amino | $9.8(2)$ | $10.1(3)$ |  |
| Phenyl/carbazate | $6.9(1)$ | $5.1(1)$ | $8.9(1)$ |
| Phenyl/S-benzyl | $82.2(1)$ | $101.6(3)$ | $100.4(3)$ |
| Carbazate/ $S$-benzyl | $83.5(1)$ | $98.8(3)$ | $96.7(2)$ |

The structure of molecule 1 was solved by direct methods. For the second molecule, the complete structure was not solved by direct methods, due to the disorder, but was completed by locating the missing atoms from successive Fourier maps. Initially, the disorder in molecule 2 led to non-convergence of refinement and unreliable bond lengths and angles. Subsequently, the $S$-benzyl group was split into two sets of atoms and their occupancies were refined at the beginning and fixed at 0.5 (the refined value), resulting in satisfactory convergence and geometric parameters. During refinement, the phenyl rings of both the disordered components of the $S$-benzyl group were restrained as regular hexagons with $\mathrm{C}-\mathrm{C}$ distances of $1.39 \AA$. All H atoms were fixed geometrically and not refined, but were allowed to ride on those atoms to which they are attached with a displacement parameter 1.2 times that of the parent atom.

Data collection: XSCANS (Siemens, 1994). Cell refinement: XSCANS. Data reduction: XSCANS. Program(s) used to solve structure: SHELXS86 (Sheldrick, 1990a). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: ORTEPII (Johnson, 1976), SHELXTL/PC (Sheldrick, 1990b). Software used to prepare material for publication: SHELXL93. Geometric calculations: PARST (Nardelli, 1983).

The authors would like to thank the Malaysian Government and the Universiti Sains Malaysia for research grant R\&D No. 123-3417-2201, and the State Science and Technology Commission, National Nature Science Foundation of China for a grant for a Key Research Project. One of the authors (KS) thanks the Universiti Sains Malaysia for a Visiting Post-Doctoral Research Fellowship.

Lists of structure factors, anisotropic displacement parameters, Hatom coordinates and complete geometry have been deposited with the IUCr (Reference: AS1178). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

## References

Ali, M. A. \& Bose, R. N. (1984). Polyhedron, 3, 517-522.
Ali, M. A. \& Tarafdar, M. T. H. (1977). J. Inorg. Nucl. Chem. 39, 1785-1791.
Davies, G., El-Sayed, M. A., El-Toukhy, A. \& Henary, M. (1990). Inorg. Chim. Acta, 168, 65-76.
Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
Nardelli, M. (1983). Comput. Chem. 7, 95-98.
Sheldrick, G. M. (1990a). Acta Cryst. A46, 467-473.
Sheldrick, G. M. (1990b). SHELXTL/PC Users Manual. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
Sheldrick, G. M. (1993). SHELXL93. Program for the Refinement of Crystal Structures. Univ. of Göttingen, Germany.
Siemens (1994). XSCANS. X-ray Single Crystal Analysis System. Version 2.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
Xu, Z., Alyea, E. C., Ferguson, G. \& Jennings, M. C. (1991). Polyhedron, 101, 1625-1629.

Acta Cryst. (1995). C51, 2083-2085

# 14-Methyl-7,14-dihydrodibenzo[a,j]acridine 

Jayanta Kumar Ray and Bddhan Chandra Roy

Department of Chemistry, Indian Institute of
Technology, Kharagpur 721 302, India
Gur Dayal Nigam
Department of Physics and Meteorology, Indian Institute of Technology, Kharagpur 721 302, India
Kandasamy Sivakumar $\dagger$ and Hoong-Kun Fun
X-ray Crystallography Laboratory, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia
(Received 3 January 1995; accepted 5 April 1995)

[^2]
[^0]:    Lists of structure factors, anisotropic displacement parameters and H-atom coordinates have been deposited with the IUCr (Reference: AS1177). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2 HU , England.

[^2]:    Abstract
    14-Methyl-7,14-dihydrodibenzo[ $a, j$ ]acridine, $\mathrm{C}_{22} \mathrm{H}_{17} \mathrm{~N}$, has butterfly-like mirror symmetry. The central ring of the acridine moiety adopts a flattened-boat conformation. The crystal structure is stabilized by van der Waals forces.

    ## Comment

    In recent years, cavity-shaped azaarenes have invaded the field of molecular recognition because of their unique structural features that help to hold the planar
    $\dagger$ On leave from: Department of Physics, Anna University, Madras 600 025, India.

